



# Graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) interfacially strengthened carbon fiber epoxy composites

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## ABSTRACT

In-situ synthesis of C<sub>3</sub>N<sub>4</sub> on the carbon fiber surface was reported for enhancing interfacial properties of carbon fiber reinforced epoxy resin composite. The formed C<sub>3</sub>N<sub>4</sub> on the carbon fiber surface can greatly increase the roughness, polar functional groups and wettability of carbon fiber surface, thereby leading to significant enhancement of interfacial properties of composites. After modification, interlaminar shear strength (ILSS) and interfacial shear strength (IFSS) of carbon fibers composites are increased from 44.3 to 60.7 MPa and from 43.1 to 75.9 MPa, respectively. Moreover, the surface free energy of carbon fibers is increased by 65.6%. The improved interfacial properties endow carbon fiber composites with better mechanical properties, leading to an increased tensile strength of composites from 1063 to 1279 MPa and total absorbed energy of impact experiment from 1.22 to 1.75 J. Meanwhile, the dynamic mechanical properties and hydrothermal aging resistance are also enhanced significantly. The storage modulus increases from 64.3 to 74.1 GPa. The markedly enhancement of interfacial mechanical properties and mechanical properties could be attributed to the improved resin wettability, enhanced mechanical interlocking and increased chemical bonding induced by the existence of C<sub>3</sub>N<sub>4</sub> on the carbon fiber surface.

## 1. Introduction

Carbon fibers (CFs) have been considered as an ideal reinforcement of advanced performance composites due to their superior mechanical properties, low weight, good stability and processing properties, outstanding thermal and electrical properties [1–3]. CFs reinforced epoxy resin materials (CFRERMs) as a potential substitute for traditional structural materials such as metal and concrete have been widely used in aerospace and space engineering, defense industry, transportation, sports, construction and other fields [4–6]. The performance and properties of CFRERMs are determined by many factors, such as reinforcement, matrix and interphase [7–10]. The interphase as a bridge between reinforcement fiber and matrix plays a crucial role in

determining the load transfer mechanisms and ultimately the physical, chemical and mechanical properties of advanced fiber-reinforced composite materials [11–16]. Unfortunately, CFs have a poor interfacial adhesion with resin matrix due to their smooth surface and chemical inertness. Poor interfacial adhesion in composites can predispose a component to debonding or delamination, thus leading to catastrophic failure [17–20]. To overcome this problem, surface modification of CF and interfacial manipulation of composites with tailored structures and properties had been extensively researched [21–26].

Up to now, various strategies including surface coating [27], chemical oxidation [28,29], ozone [30–32], plasma treatment [33], high energy irradiation [34] and nanomaterials [35,36] have been developed to improve fiber/matrix adhesion. Among them, an effective

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method to ameliorate the interfacial quality of CFRERMs is developing hierarchical (or multiscale) micro/nano reinforcements where nanoscale materials are integrated with CFs [37,38]. In this approach, nanomaterials are grafted or anchored onto the surface of CFs under the action of physical and/or chemical adsorption. These nanomaterials (such as graphene, carbon nanotubes (CNTs), nanoclay, metal oxide nanoparticles, etc.) endow interphase with enlarged contact area, intensified mechanical interlocking, strong chemical bonding and/or local stiffening between the fiber and matrix [39–41]. All these factors benefit improving the stress transfer from strong fillers to the polymer matrix and interfacial properties of the final composites.

Graphitic-based nanomaterials such as CNTs, graphene and graphene oxide were widely used to fabricate hierarchical CFs using different techniques [42,43]. The easiest approach to coat graphitic-based materials onto the CF is surface coating (dip coating and spray coating). This surface coating can effectively protect the fibers against damage or friction, improve the interfacial properties and out-of-plane shear properties of final composites. For example, Zhang et al. [44] deposited CNT networks on CF preregs using spray coating to improve out-of-plane mechanical properties, and found that Mode-I fracture toughness of the CF laminates was increased by 50% in the presence of 0.047 wt% CNTs. Yao et al. [45] modified CF with CNTs for improving interfacial properties of CF/epoxy composites in a sizing deposition process. The formed gradient interfacial structure in the CF/epoxy composites helped transfer the stress uniformly and made the interlaminar shear strength (ILSS) increased by 13.45%.

Many studies have clearly indicated that the creation of hierarchical CF through CVD is a very straightforward approach for interfacial manipulation [3,4]. In this technique, CNTs are directly grown on the CF surface at high temperature in the presence of catalyst. This results in the improvement of interfacial interlocking, and effective stress transfer between the fibers and matrix [11,46]. However, the high temperature in CVD processes easily led to the deterioration in the tensile properties of CFs, and some catalysts and carbon source were costly and toxic [47]. Qian et al. [42] uniformly grafted CNTs onto Hexcel® IM7 CFs using a CVD method for a better stress transfer between fibers and matrix and the interfacial shear strength (IFSS) was increased by 26%. Meanwhile, an obviously degraded tensile strength of CFs (around 15%) resulted from the observed dissolution of catalyst into the CFs.

Moreover, electrophoretic deposition (EPD) is also a practical and cost-efficient method to fabricate hierarchical composites [38,43]. With the help of EPD, CNTs, graphene and aramid nanofibers (ANFs) were successfully coated on the CF surface and thus improved the surface energy and IFSS. Schaefer et al. [48] deposited carboxylic acid-functionalized carbon nanofibers (O-CNFs) on the surface of single CFs, and these O-CNFs coated CFs provided an IFSS increase of 207.6%. Rodriguez et al. [49] also prepared the hierarchical fabric by depositing oxidized CNFs and amidized CNFs on the surface of CF layers. The relevant composites containing oxidized CNFs or amidized CNFs showed an increase in ILSS of 9.08% and 12.44%, respectively. Sui et al. [50] built a transition layer of oxidized MWCNTs on CFs using a continuous EPD method. The static and fatigue tests showed an increase of 33.3% in IFSS, 10.5 in ILSS, 9.5% in flexural strength, and 15.4% in flexural modulus, respectively.

Chemical reaction is another common strategy to graft graphitic-based nanomaterials onto CFs surface for improved interfacial properties [46,51–53]. In the pioneering research of Islam et al. [54], CNTs were directly grafted onto CFs through ester linkage in the absence of catalyst or coupling agents. Pulling out experiment confirmed the existence of strong carbon-carbon bonding in the interphase, which provided significantly improved interfacial and impact properties. For example, Zhao et al. [51] fabricated a CF/polyhedral oligomeric silsesquioxane (POSS)/CNT hierarchical reinforcement using octaglycidyl dimethylsilyl POSS as the linkage between CFs and CNTs. Both the ILSS and impact resistance of final hierarchical composites had a

dramatic improvement compared with the composites containing untreated CFs. In another study, He et al. [55] fabricated CNT-CF hierarchical structures using poly-(amidoamine) as a “bridging” in a chemical reaction process. The directly measured grafting strength between individual CNT and CFs substrate varied between the van der Waals (5 MPa) force and 7 times of van der Waals force (90 MPa).

Although numerous methods have been devised for further development of hierarchical reinforcements. But the application of these hierarchical reinforcements is frequently limited by complicated technology and expensive reagents. Therefore, the greatest challenge of hierarchical reinforcements is still searching for a facile and effective approach for improved interfacial properties. In the present work, a facile and effective strategy to fabricate hierarchical carbon fibers is reported for enhanced interfacial properties of CFRERMs through in-situ synthesis of graphitic carbon nitride ( $g\text{-C}_3\text{N}_4$ ) on the CF surface. The enhanced effect of  $g\text{-C}_3\text{N}_4$  on interfacial adhesion is superior to those of the other carbon fiber/epoxy composites (as shown in Table S2). Moreover, surface morphology of CFs and fracture appearance of CFs composites were detected by scanning electron microscope (SEM). Functional groups and chemical elements of CFs and  $g\text{-C}_3\text{N}_4$  were examined by Fourier Transform Infrared Spectroscopy (FTIR) and X-ray photoelectron spectroscopy (XPS), respectively. Wettability studies were performed on a dynamic contact angle analysis (DCAT). The interfacial properties of CFs composites were evaluated by interlaminar shear strength (ILSS), interfacial shear strength (IFSS), and hydrothermal aging resistance. Meanwhile, the practical application of CFs composites was simulated by dynamic thermomechanical analysis (DMA). As far as we know, there are no relevant research about  $g\text{-C}_3\text{N}_4$  as reinforcement for improved interfacial properties of fiber/resin composites.

## 2. Results and discussion

### 2.1. Surface morphology of carbon fibers

Fig. 1 shows the morphology change of CFs induced by in-situ synthesized  $g\text{-C}_3\text{N}_4$ . Some obvious alterations on the surface morphology were observed. A smooth and neat surface can be found on CFs, and some grooves are uniformly distributed parallel to the fiber axis on the CF surface. After modification,  $g\text{-C}_3\text{N}_4$  was successfully synthesized and uniformly distributed on the CFs- $\text{C}_3\text{N}_4$  surface, which led to the rougher surface.

### 2.2. Surface chemical elemental composition of carbon fibers

The surface chemical compositions of CFs,  $\text{C}_3\text{N}_4$  and CFs- $\text{C}_3\text{N}_4$  were analyzed by XPS. As illustrated in Table 1, the surface of as-prepared samples contains carbon, nitrogen and oxygen element. The introduction of  $\text{C}_3\text{N}_4$  causes a sharp increase in nitrogen content on the CFs- $\text{C}_3\text{N}_4$  surface. The XPS C1s spectra of samples (Fig. 2) show only a characteristic peak at about 284.6 eV for the CFs which is attributed to the  $\text{sp}^2$  hybridized C atoms in the C–C group [56–59]. Meanwhile, pure  $\text{C}_3\text{N}_4$  exhibits two characteristic peaks at 284.6 and 288.1 eV, respectively. The characteristic peak at 288.1 eV is assigned to the  $\text{sp}^2$ -bonded C in  $\text{N}=\text{C}(\text{N})_2$  [60]. After surface modification, XPS C1s spectra of CFs- $\text{C}_3\text{N}_4$  show a significant characteristic peak of  $\text{sp}^2$ -bonded C in the  $\text{N}=\text{C}(\text{N})_2$  group. These results of XPS analysis confirm the formation of  $\text{C}_3\text{N}_4$  on the CFs.

The surface functional groups of CFs and  $\text{C}_3\text{N}_4$  have also been studied by FTIR analysis. As illustrated in Fig. 3, no significant characteristic absorption peak was observed on CFs. After coated with urea, some characteristic absorption peaks of functional groups of urea can be found. The absorption peaks in the range of  $3600\text{--}3100\text{ cm}^{-1}$  and  $900\text{--}650\text{ cm}^{-1}$  represent the stretching vibration and in-plane bending vibration of N–H bond, respectively. The characteristic absorption peaks at  $1680\text{--}1630\text{ cm}^{-1}$  reflect the stretching vibration of the C=O

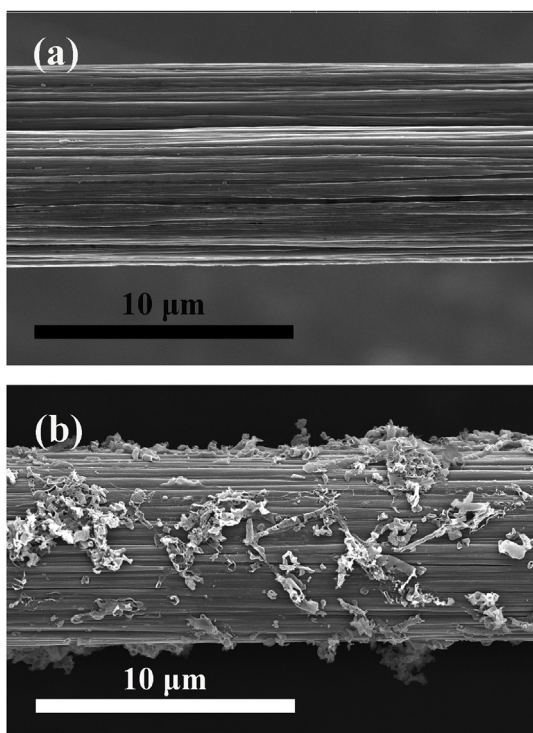


Fig. 1. Surface morphology of carbon fibers: SEM microstructure of (a) CFs; and (b) CFs-C<sub>3</sub>N<sub>4</sub>.

Table 1  
Surface chemical composition of desizing CFs with and without C<sub>3</sub>N<sub>4</sub>.

Carbon fiber	Element content (%)				
	C	O	N	O/C	N/C
Desizing CFs	83.42	15.27	1.31	18.30	1.57
C <sub>3</sub> N <sub>4</sub>	43.81	8.05	48.14	18.37	109.88
CFs-C <sub>3</sub> N <sub>4</sub>	55.21	6.84	37.95	12.39	68.74

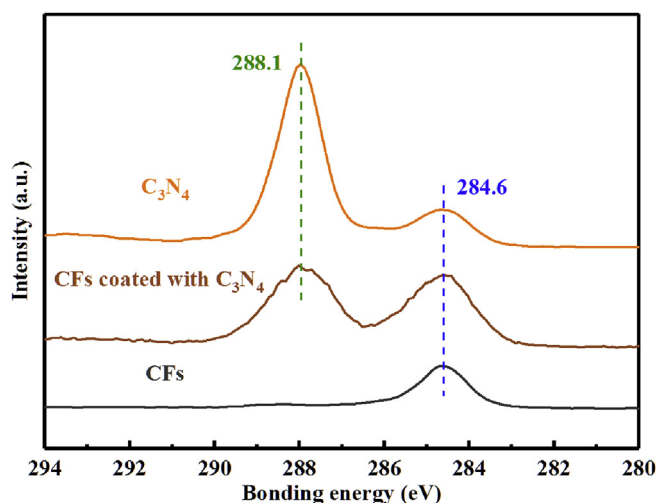


Fig. 2. XPS C1s spectra of CFs, C<sub>3</sub>N<sub>4</sub> and CFs-C<sub>3</sub>N<sub>4</sub>.

bond, and the absorption peaks at 1450 and 1160 cm<sup>-1</sup> are attributed to the stretching mode and vibration mode of the C–N bond, respectively. For pure C<sub>3</sub>N<sub>4</sub>, the absorption peaks at around 3500–3000 cm<sup>-1</sup> were ascribed to the vibration mode of N–H bond and to the stretching mode of O–H bond. The absorption peak at 2178 and 806 cm<sup>-1</sup> can be

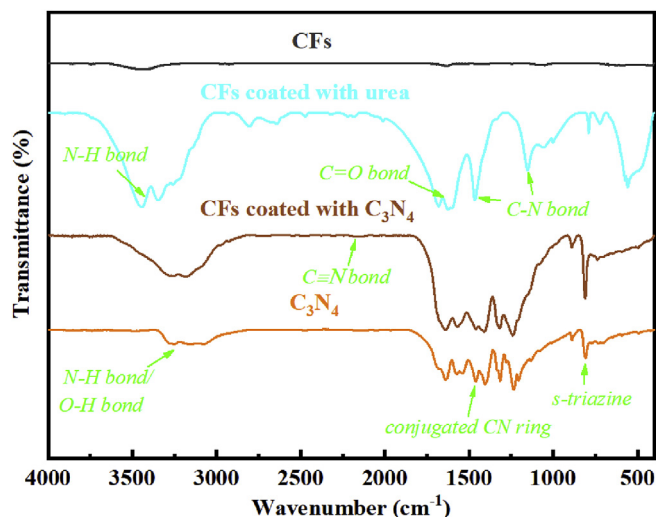


Fig. 3. FTIR spectra of carbon fibers and C<sub>3</sub>N<sub>4</sub>.

attributed to the C≡N triple bonds and breathing mode of s-triazine, respectively [60]. The absorption peaks in the range of 1700–1200 cm<sup>-1</sup> were assigned to the stretching vibration of conjugated CN rings. In comparison with C<sub>3</sub>N<sub>4</sub>, the CFs-C<sub>3</sub>N<sub>4</sub> sample showed a similar absorption in the wavenumber range of 4000–400 cm<sup>-1</sup>. The in-situ synthesis of C<sub>3</sub>N<sub>4</sub> on CFs can effectively increase the surface functional groups and surface polarity of CFs, thus ameliorate the interfacial wettability and enhance the interfacial adhesion between reinforcement and matrix.

The formation of C<sub>3</sub>N<sub>4</sub> on CFs was also confirmed by TGA. Fig. 4 shows the TGA curves. For the CFs sample, a weight loss of 3.72% was observed in the temperature range of 50–800 °C due to thermal degradation of surface materials on the CFs. The TGA curve of CFs-C<sub>3</sub>N<sub>4</sub> is similar to that of CFs in the temperature range of 50–500 °C. However, a significant difference between CFs and CFs-C<sub>3</sub>N<sub>4</sub> occurred in the temperature range of 500–800 °C owing to the decomposition of C<sub>3</sub>N<sub>4</sub>. Therefore, CFs-C<sub>3</sub>N<sub>4</sub> has a higher weight-loss ratio (4.25%). The above TGA results indicated that the C<sub>3</sub>N<sub>4</sub> was synthesized on the CF surface, which was consistent with the XPS and FTIR analysis.

### 2.3. Dynamic contact angle analysis

The chemical state and topography of the CFs surface influence its surface energy. The high surface energy represents a better wettability

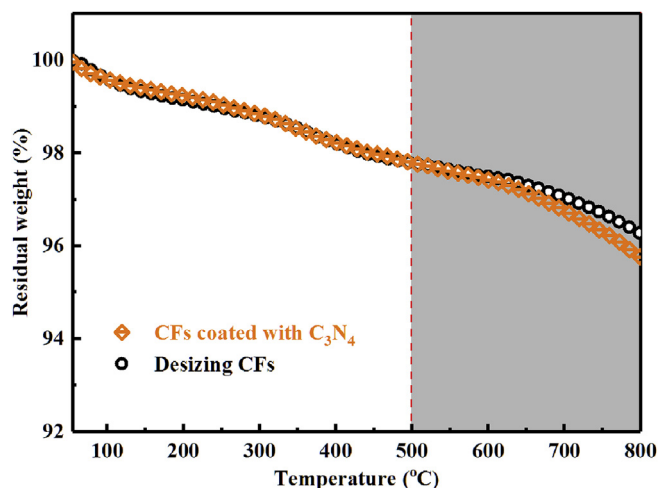


Fig. 4. TGA curves of CFs and CFs-C<sub>3</sub>N<sub>4</sub>.

**Table 2**  
Contact angle, surface energy and roughness of desizing CFs with and without C<sub>3</sub>N<sub>4</sub>.

Carbon fibers	Contact angle (°)		Surface energy (mJ/m)		
	Deionized water	Diiodomethane	$\gamma^d$	$\gamma^p$	$\gamma$
Desizing CFs	75.37	59.68	28.75	8.28	37.03
CFs-C <sub>3</sub> N <sub>4</sub>	46.37	34.45	42.28	19.03	61.31

between CFs and matrix, and strong interfacial adhesion. The surface wettability of CFs was investigated by a Cahn dynamic angle analysis system. Contact angle ( $\theta$ ), surface energy ( $\gamma$ ) and its sub-component: dispersion component ( $\gamma^d$ ) and polar component ( $\gamma^p$ ) of CFs and CFs-C<sub>3</sub>N<sub>4</sub> were summarized in Table 2. Compared with CFs, CFs-C<sub>3</sub>N<sub>4</sub> possesses smaller contact angles and larger surface energy. The introduced C<sub>3</sub>N<sub>4</sub> on the CFs surface causes the contact angle of deionized water (polar solvent) reduced from 75.37° to 46.37°, the contact angle of diiodomethane (non-polar solvent) is decreased from 59.68° to 34.45°, and the surface energy of CFs-C<sub>3</sub>N<sub>4</sub> is increased from 37.03 to 61.31 mJ/m. The increased  $\gamma^d$  (from 28.75 to 42.28 mJ/m) and  $\gamma^p$  (from 8.28 to 19.03 mJ/m) could be ascribed to the sharp increase in the surface roughness and functional groups (-NH<sub>2</sub>) induced by C<sub>3</sub>N<sub>4</sub>. Therefore, the hierarchical structure of CFs-C<sub>3</sub>N<sub>4</sub> with high surface energy can effectively improve the interfacial wettability between CFs and epoxy resin matrix, thus markedly enhance the interfacial properties (listed in 2.4 and 2.6 section) of the CF composites.

**2.4. Interfacial properties evaluation**

The interfacial properties has a major impact on the mechanical properties of the final composites. Here, the ILSS as a vital factor of the interface strength of composite laminates was evaluated by short beam bending test (Fig. 5). The ILSS of CFs composites and CFs-C<sub>3</sub>N<sub>4</sub> composites is 44.3 and 60.7 MPa, respectively. The introduced C<sub>3</sub>N<sub>4</sub> on the CFs surface makes ILSS increase by about 37.2%. The notably enhanced interfacial strength can be attributed to enhanced mechanical interlocking and strong chemical bonding in the interphase. The introduced C<sub>3</sub>N<sub>4</sub> endows interphase with bigger contact area, better surface wettability and stronger mechanical interlocking. Meanwhile, open-ring reaction between amino groups on C<sub>3</sub>N<sub>4</sub> surface with epoxy groups of matrix creates sufficient chemical bonding between reinforcement and matrix.

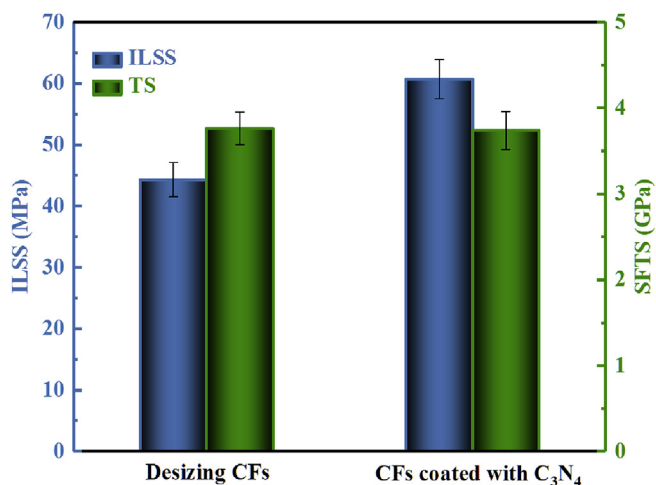


Fig. 5. Tensile strength (TS) of single fiber and ILSS.

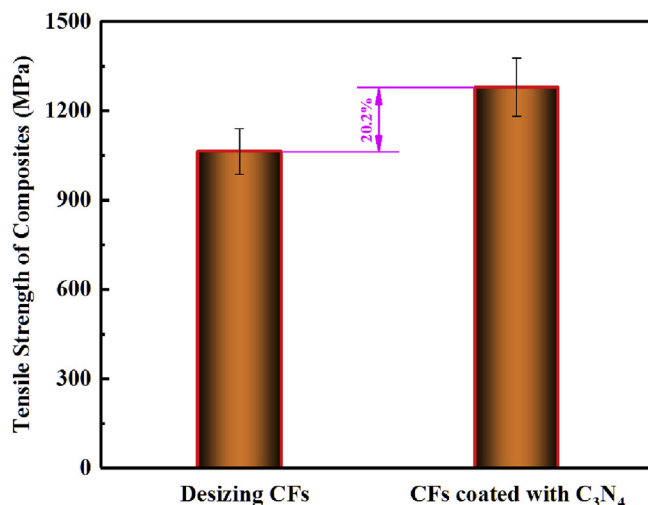


Fig. 6. Tensile strength of composites reinforced with CFs and CFs-C<sub>3</sub>N<sub>4</sub>, respectively.

**2.5. Mechanical properties**

The tensile strength (TS) of single fiber was investigated for revealing the effects of surface modification on the inherent mechanical properties of CFs. As illustrated in Fig. 5, the single fiber TS (SFTS) of CFs and CFs-C<sub>3</sub>N<sub>4</sub> is 3.76 and 3.74 GPa, respectively. The results of SFTS suggest that the in-situ synthesis of C<sub>3</sub>N<sub>4</sub> has no significant effects on the inherent mechanical property of CFs. Moreover, the TS of composites was also studied by unidirectional tensile test of CFs composites and CFs-C<sub>3</sub>N<sub>4</sub> composites. As Fig. 6 shows, the C<sub>3</sub>N<sub>4</sub> anchored on CFs surface greatly enhances the TS of CFs-C<sub>3</sub>N<sub>4</sub> composites. The TS increases from 1063 to 1279 MPa. The markedly improved TS (increased by 20.2%) can be ascribed to the synergistic effects of enhanced mechanical interlocking, wettability and chemical bonding between CFs and epoxy matrix [20]. Meanwhile, the results of mechanical properties test confirm that CFs-C<sub>3</sub>N<sub>4</sub> composites have better mechanical properties than those of the as-received CFs composites (as shown in Table S1).

Impact properties as a major embodiment of interfacial property of composites reinforced with CFs and CFs-C<sub>3</sub>N<sub>4</sub> were explored. Fig. 7 shows the corresponding results. The initial, propagative and total absorbed energy of CFs composite were 0.24, 0.98 and 1.22 J,

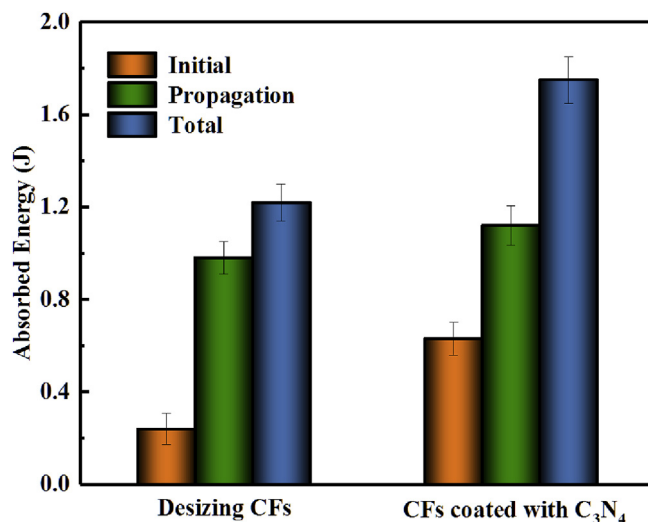


Fig. 7. Impact properties of composites reinforced with CFs and CFs-C<sub>3</sub>N<sub>4</sub>, respectively.

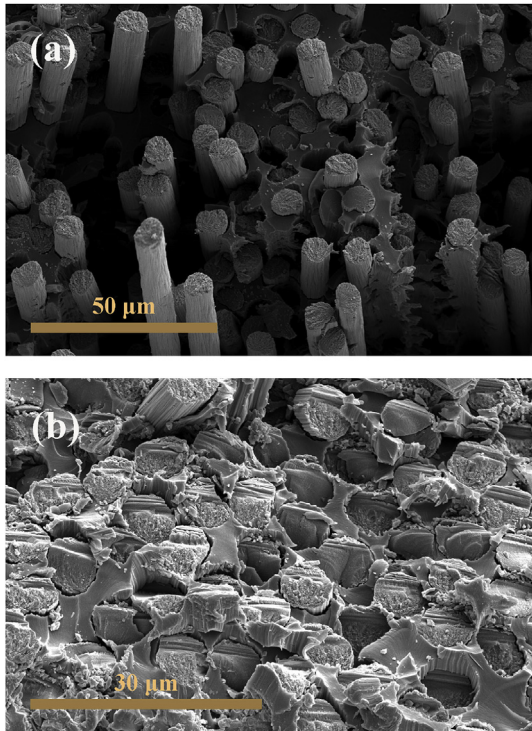


Fig. 8. SEM micrographs of typical failure surfaces for CFs-epoxy composites (a) CFs; and (b) CFs coated with C<sub>3</sub>N<sub>4</sub>.

respectively. The in-situ formed C<sub>3</sub>N<sub>4</sub> on CFs surface effectively enhanced the impact properties of CFs-C<sub>3</sub>N<sub>4</sub> composites. The initial, propagative and total absorbed energy of CFs-C<sub>3</sub>N<sub>4</sub> composite increased to 0.63, 1.12 and 1.75 J, respectively. The interphase of composites acts as a shielding layer to prevent the crack tips to directly contact the CFs by triggering sub-cracks to absorb the fracture energy. Clearly, the complex interphase structure created by C<sub>3</sub>N<sub>4</sub> can trigger more sub-cracks and absorbs more fracture energy, thereby efficiently enhances the impact failure resistance. Meanwhile, the impact fracture surface of CFs composites and CFs-C<sub>3</sub>N<sub>4</sub> composites was characterized by SEM, as shown in Fig. 8. For CFs composites, interfacial de-bonding, cracks and pull-out of fiber induced by poor interfacial adhesion are observed on the fracture surface and no resin matrix can be found on the fiber surface. In the case of CFs-C<sub>3</sub>N<sub>4</sub> composites, the flat fracture appearance implies a strong and powerful interfacial adhesion created by enhanced mechanical interlocking, better wettability and improved chemical bonding exists in the interfacial area between CFs and epoxy resin matrix.

Fig. 9 depicts the schematic of the impact test model to reveal the

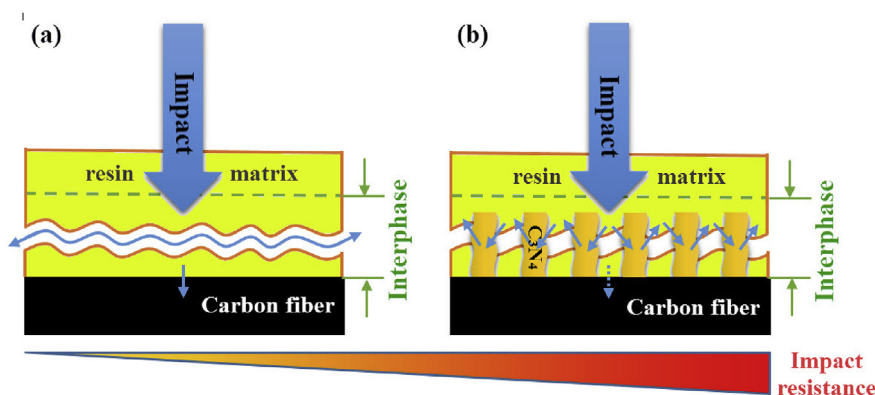


Fig. 9. Schematic impact test of epoxy composites reinforced with CFs and (b) CFs-C<sub>3</sub>N<sub>4</sub>.

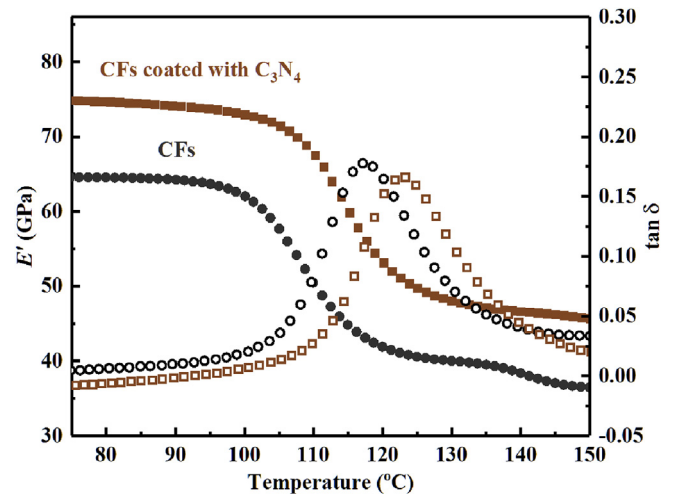


Fig. 10. Dynamic mechanical analysis (DMA) of CFs composites and CFs-C<sub>3</sub>N<sub>4</sub> composites.

mechanism of interphase in the composites. The inferior interfacial bonding of CFs composites can result in an ineffective load transfer between CFs and epoxy resin, and fast propagation of cracks along the interphase under a low impact load. Compared to CFs composites, the strong and complicated interfacial structure of CFs-C<sub>3</sub>N<sub>4</sub> composites derived from enhanced interfacial chemical bonding and interfacial mechanical interlocking force can markedly improve the interfacial load transfer and release the stress concentration. Meanwhile, the complicated propagation of cracks and enhanced deformation ability of resin led to the formation of micron cracks and change of failure mode. The complex propagation mode of crack consumes more energy, which results in the significantly increased impact properties.

In addition, the dynamic mechanical properties were also studied to simulate the practical applications of composites. Fig. 10 displays the variation of storage modulus ( $E'$ ) and mechanical loss ( $\tan \delta$ ) with temperature for the CFs composites and CFs-C<sub>3</sub>N<sub>4</sub> composites. Compared with CFs composites, CFs-C<sub>3</sub>N<sub>4</sub> composites have a higher  $E'$ . The presence of C<sub>3</sub>N<sub>4</sub> on the CFs enables  $E'$  to increase from 64.3 to 74.1 GPa below  $T_g$  (glass transition temperature) and from 36.8 to 45.9 GPa above  $T_g$ , which means a better stability of stiffness at higher temperature near  $T_g$ . The enhancement of  $E'$  was attributed to the higher volume fraction of interphase in the composites and lower mobility of polymer chains at the interphase region. Therefore, the interphase can be considered as additional reinforcement for mechanical stiffening of the composites. Similar phenomenon was also found in the CNTs/CFs composites [39]. Furthermore, the  $\tan \delta$  curves illustrates that CFs-C<sub>3</sub>N<sub>4</sub> composites have a higher  $T_g$  (122.9 °C) and lower  $\tan \delta$  compared with that of CFs composites (117.4 °C). The increased  $T_g$

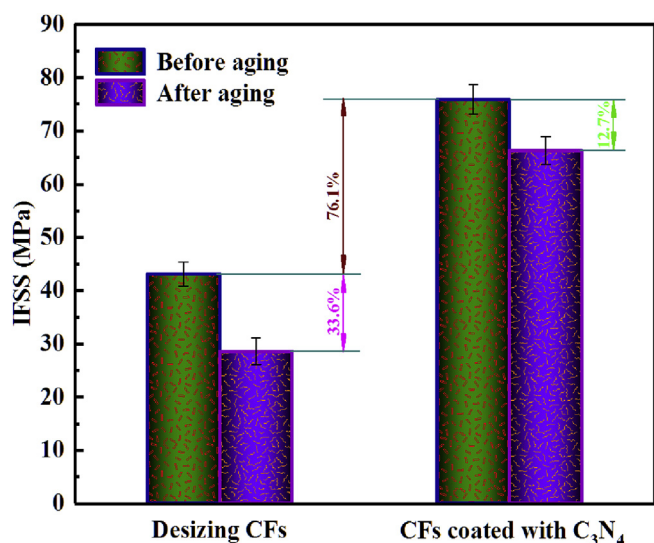


Fig. 11. IFSS before and after hydrothermal aging for 48 h.

could be related to the enhanced restricted-mobility of polymer chains in the interphase region [39]. The amino groups on C<sub>3</sub>N<sub>4</sub> can ameliorate the wettability and reaction between fibers and matrix, thereby enhances the crosslink density of interphase [37]. In addition, the C<sub>3</sub>N<sub>4</sub> can impede the viscous flow of polymer chains and minimize the energy loss of viscous deformation.

### 2.6. Hydrothermal aging resistance

High humidity environment can degrade the mechanical properties of composites by the hydrolysis and damage of the interphase structure. Here, the hydrothermal aging resistance of composites reinforced with CFs and CFs-C<sub>3</sub>N<sub>4</sub> is also evaluated (Fig. 11). It can be found that the formation of C<sub>3</sub>N<sub>4</sub> on CFs results in the IFSS of composite increasing from 43.1 to 75.9 MPa with a growth rate of 76.1%. After hydrothermal aging treatment, the IFSS of CFs and CFs-C<sub>3</sub>N<sub>4</sub> composite is reduced by 33.6% and 12.7%, respectively. The interfacial adhesion is poor due to the absence of chemical bond existing in the interphase, thus water molecules can easily penetrate into the interphase, seriously destroys the interfacial structure and obviously degrades the interfacial properties. The enhanced hydrothermal aging resistance of CFs-C<sub>3</sub>N<sub>4</sub> composite can be attributed to the stronger interfacial adhesion resulting from enhanced chemical bonds and mechanical interlocking between reinforcements and matrix. Moreover, the damage of C–N bonds formed by opening-ring reaction between amino groups on C<sub>3</sub>N<sub>4</sub> surface and epoxy groups of matrix needs more energy, thus protect the interphase from destruction efficiently.

### 3. Conclusions

g-C<sub>3</sub>N<sub>4</sub> was in-situ synthesized on carbon fiber surface for the improvement of interfacial properties between CFs and epoxy matrix. The introduced g-C<sub>3</sub>N<sub>4</sub> results in the marked increase of roughness, polar functional groups and surface free energy of CFs surface. The surface free energy of CFs was increased from 37.03 to 61.31 mJ/m. The interfacial properties of carbon fibers composites were significantly improved after modification. The ILSS and IFSS of composites were increased from 44.3 to 60.7 MPa and from 43.1 to 75.9 MPa, respectively. The excellent interfacial properties endowed CFs composites with outstanding mechanical properties. The tensile strength of composites was increased from 1063 to 1279 MPa and total absorbed energy of impact experiment was increased from 1.22 to 1.75 J. In addition, the dynamic mechanical properties and hydrothermal aging resistance were also enhanced significantly. The  $E'$  of the CFs composites was increased from

64.3 to 74.1 GPa below  $T_g$  and from 36.8 to 45.9 GPa above  $T_g$ . The hydrothermal aging test demonstrated that the reduction rate of IFSS of CFs composites decreases from 33.6 to 12.7%. The enhanced interfacial properties and mechanical properties of CFs-C<sub>3</sub>N<sub>4</sub> were attributed to the improved wettability, enhanced mechanical interlocking and increased chemical bonding. The reported surface modification used in this study can markedly improve the interfacial properties and mechanical properties of carbon fiber composites for other applications like anticorrosion coating and energy storage/conversion units [61–64].

### Declaration of interest

The authors declare no conflict of interest regarding the publication of this paper.

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### Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.compscitech.2018.08.031>.

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